Novel tricyclic spiropiperidines. Synthesis and adrenergic activity of spiro[1,3-benzodioxolopiperidines] and spiro[1,3-benzodioxanepiperidines]

MD Pujol1* G Rosell1, G Guillaumet2

¹Laboratori de Química Farmacèutica, Facultat de Farmàcia, Universitat de Barcelona, Av Diagonal 643, 08028 Barcelona, Spain; ²Institut de chimie organique et analytique, Université d' Orléans, BP 6759, 45067 Orléans Cedex 2, France

(Received 15 November 1995; accepted 6 May 1996)

Summary — The synthesis of new series of spiropiperidines is reported. Some of these compounds showed interesting adrenergic activity. The biological activities of the new compounds were evaluated on-guinea pig atria (β 1), guinea-pig trachea (β 2) and rat vas deferens (α). Compounds **6a**, **6c** and **6f** showed activity comparable to propranolol, in spite of being tertiary amines.

spiropiperidine / adrenergic antagonist / tertiary amine

Introduction

The synthesis and pharmacological activity of a series of spiropiperidines exemplified by benzofuran derivatives 1 and isobenzofuran 2 (fig 1) has been described previously [1, 2].

The heterocyclic system has served as a source for a variety of pharmacodynamic agents. Noteworthy antidepressant [2], neuroleptic [3], hypotensive [4] and diuretic [5] activities have been demonstrated in animal models by various compounds with the spiropiperidine subunit in their structure but are unknown as adrenergic blockers. It is known that aryloxypropanolamines are the most important structural class of β -adrenergic antagonists [6]. Nevertheless, structureactivity relationships have not been completely established and systematic studies indicated that the tertiary amine in this structure reduces the potency markedly [7].

We decided to synthesize a series of new tricyclic 2,4'-spiro-1,3-benzodioxolo and 2,4'-spiro-1,3-benzodioxanepiperidines combined with several types of *N*-substituents (compounds **6a-0**), in order to ascertain their activity as a class of adrenergic compounds (fig 2).

$$R_{1} + C_{0} + C_{0$$

Fig 1. Structures of compounds 1 and 2.

Chemistry

The synthetic route used to prepare compounds 6a-o is illustrated in scheme 1. 1,3-Benzodioxole derivatives have considerable importance in biological chemistry and their syntheses have been investigated on several occasions [8]. It was shown that the reaction of acetalization of catechol with ketones was efficiently catalysed by a trace of p-toluensulfonic acid (pTSA) [9], and the water thus formed could readily be removed by azeotropic distillation with benzene or toluene. However, the procedure was not satisfactory in our work. Under these conditions the starting material was recovered, even after being refluxed for a long period of time. Our modified procedure was the transacetalization of ketone dimethyl acetal, obtained from the N-acetyl-4-piperidone and methyl orthoformate [10], with 1,2-di-

^{*}E-mail: mdpujol@farmacia. far.ub.es

$$R_{2}=\mathbf{u}: \ \ \mathbf{CH}_{2}\cdot\mathbf{CH}(\mathrm{OH})\cdot\mathbf{CH}_{2}\cdot\mathbf{O} \longrightarrow \mathbf{NHCOCH}_{3} \qquad R_{2}=\mathbf{t}: \qquad \begin{array}{c} \mathbf{OH}_{2}\cdot\mathbf{H} \\ \mathbf{OH}_{2}\cdot\mathbf{CH} \\ \mathbf{R}_{2}=\mathbf{r}: \ \ \mathbf{CH}_{2}\cdot\mathbf{CH}(\mathrm{OH})\cdot\mathbf{CH}_{2}\cdot\mathbf{O} \longrightarrow \mathbf{R}_{2}=\mathbf{r}: \\ \mathbf{R}_{2}=\mathbf{r}: \ \ \mathbf{CH}_{2}\cdot\mathbf{CH}_{2} \longrightarrow \mathbf{CH}_{2}\cdot\mathbf{CH}_{2} \longrightarrow \mathbf{CH}_{2} \longrightarrow \mathbf{C$$

Series	Compound	n	R_I	R_2
I	6a	0	Н	u
	6b	0	Н	r
	6c	0	Н	s
	6d	0	Н	t
	6e	0	H	v
II	6 f	1	Н	u
	6g 6h	1	Н	r
	6 ň	1	Н	S
	6i	1	Н	t
	6j	1	Н	V
Ш	6k	0	Br	u
	6 l	0	Br	r
	6m	Ō	Br	S
	6n	0	Br	t
	60	0	Br	v

Fig 2. Structures of compounds 6a-o.

hydroxybenzene. The same method was employed to obtain the 1,3-benzodioxane derivatives from 2-hydroxybenzyl alcohol (scheme 1). Compound **5c** was directly obtained by treatment of hydrochloride **5a** with *N*-bromosuccinimide (NBS) in the methanol at room temperature.

Acetals derived from N-acetyl-4-piperidone were, upon hydrolysis, converted to 6a—o by reaction with alkyl iodide (7e) or from the cleavage of the respective epoxides (7a—d). The intermediate compounds 7a,b were prepared in good yields by reaction of epichlorohydrin with the appropriate phenol in basic media [11] (scheme 2). The *erythro* and *threo* 7c,d were obtained from the reactions between catechol and the *trans*- or *cis*-2,3-bis(chloromethyl)oxirane respectively [12] (scheme 3). The reduction of ester with LiAlH₄ afforded compound 7e [13]. The hydroxy group was then displaced by the iodine atom via the Mitsunobu reaction [14, 15] (scheme 4).

Scheme 1. Reagents and conditions: a) MeOH/pTSA; b) 1,2-dihydroxybenzene or 2-hydroxybenzyl alcohol; c) KOH/H₂O; d) (i) HCl/ether, (ii) NBS/MeOH; e or f) 7a–7e, DMF, K₂CO₃, stirring, rt.

Pharmacology

The biological profiles of the compounds listed in table I at α , β_1 and β_2 adrenoceptors were assessed on rat vas deferens, electrically stimulated guinea-pig left atria and guinea-pig tracheal chains, respectively. Isoproterenol was used as the agonist; detailed test procedures have been reported previously [16]. The potency of drugs is expressed as pA_2 calculated according to Van Rossum [17].

Results and discussion

These compounds **6a–o** were evaluated in vitro for their α -blocking activity against noradrenaline in rat vas deferens. The in vitro β -blocking potency of **6a–o** was measured against isoprenaline, in tracheal smooth muscle (β_2) and electrically stimulated left atrium (β_1) of guinea pigs. Compounds **8** and **9** are aryloxypropanolamine analogues of propranolol and have a tertiary amine as the *N*-substituent (piperidine or 4-piperidone); these were prepared by classical methods [11,

Scheme 2.

18] and were tested under the same conditions and compared with compounds **6a–o**. The resulting potencies are collected in table I.

In the present study, the potency of each series (I, II and III) was generally a function of the *N*-substituents.

Compounds **6e**, **6j** and **6o**, with a substituent without hydroxyl group on the *N*-substituent, showed β -adrenergic inactivity and little α -blocking activity. This result is compatible with those obtained for Piperoxan and WB-4101, which are known as α -blocking compounds [19].

Scheme 3.

$$\begin{array}{c}
COOE_1 \\
\hline
COOE_2 \\
\hline
CH_2I
\end{array}$$

$$\begin{array}{c}
CH_2I \\
\hline
CH_2I
\end{array}$$

Scheme 4.

The higher mean pA_2 on atrium found for compounds **6a** and **6f** indicated that the aryloxy-propanolamine subunit and the aromatic ring with the acetamido substituent in position 4 are the best groups in these spiro series for adrenergic activity.

Table I. Activity values for compounds 6a-o.

Compound	β-Adrenergic antagonisma		β_1/β_2	α-Adrenergic antagonism	
	$pA_2(\beta I)^c$	$pA_2 (\beta_2)^{\mathrm{d}}$		<i>pD'</i> ₂ e	
6a	8.42 ± 0.31	7.59 ± 0.25	6.8	Inactive	
6b	7.21 ± 0.12	7.14 ± 0.28	1.2	Inactive	
6c	8.15 ± 0.15	6.96 ± 0.21	15.5	Inactive	
6d	< 5	< 5		Inactive	
6e	Inactive	Inactive		5.57 ± 0.22	
6f	8.35 ± 0.24	7.23 ± 0.35	13.2	Inactive	
6g	7.75 ± 0.19	7.70 ± 0.17	1.1	Inactive	
6h	8.07 ± 0.23	7.04 ± 0.15	10.7	Inactive	
6i	5.60 ± 0.27	4.90 ± 0.35	5.0	Inactive	
6 j	Inactive	Inactive		4.27 ± 0.21	
6k	7.57 ± 0.28	7.12 ± 0.36	2.9	Inactive	
61	6.23 ± 0.15	6.75 ± 0.22	0.3	Inactive	
6m	7.80 ± 0.12	7.32 ± 0.37	3.0	Inactive	
6n	< 5	< 5		Inactive	
60	Inactive	Inactive		5.01 ± 0.21	
Propranolol	8.60 ± 0.32	8.47 ± 0.25	1.4	Inactive	
8	7.69 ± 0.33	7.52 ± 0.18	_	Inactive	
9	7.04 ± 0.35	6.84 ± 0.13	_	Inactive	

 $^{a}pA_{2}$ values \pm SD were calculated in the same way as described by Van Rossum [17]. A minimum of three different concentrations were tested in each preparation. Where pA_{2} values are given, the slope of Schild plot was equal to 1 ± 0.1 . b Non-competitive antagonism in vas deferens is expressed as the pD'_{2} value. c Antagonism of the isoprenaline-induced positive inotropic effect in guinea-pig left atrium. d Antagonism of the isoprenaline-induced relaxation effect in tracheal preparation. c Antagonism of the noradrenaline-induced contraction of rat vas deferens.

In previous studies on adrenergic compounds, the conversion of secondary amine into tertiary amine decreased the activity, as in the case of compounds 8 and 9 (fig 3) in comparison with the propranolol [20]. Tertiary amines were not markedly active as α - or β -adrenergic blocking agents, but had some anticonvulsive activity on the central nervous system [21], whereas these series of compounds 6a–0 with the N-cyclic group are less affected by structural modifications than classic aryloxypropanolamines.

Fig 3. Structures of compounds 8 and 9.

The *erythro* isomers with 1-(1,4-benzodioxan-2-yl)ethanol as the *N*-substituent were consistently more active than their *threo* counterparts, a result which agrees with literature data for α -branched β -blockers [16] or β -agonists [22–24]. The *erythro* compound **6c** is the most active β_1 -blocker; with this stereochemistry its selectivity β_1/β_2 is ten times more than the propranolol and its very potent competitive β_1 -adrenergic antagonism is potentially the best in the series.

In summary, this study shows that when some analogues were substituted on the basic nitrogen by specific aralkyl groups, they showed α -blocking activity ($6\mathbf{e}$, \mathbf{j} , \mathbf{o}) or β -blocking activity ($6\mathbf{a}$ - \mathbf{c} , \mathbf{f} - \mathbf{i} , \mathbf{k} - \mathbf{m}). It is interesting to note that in the cases $6\mathbf{a}$, $6\mathbf{c}$, $6\mathbf{f}$ and $6\mathbf{h}$ high β_1 -activity is observed, which is similar the propranolol activity. But there are no compounds known with a combination of effects. Substitution in the aromatic ring ($6\mathbf{k}$, $6\mathbf{l}$ and $6\mathbf{m}$) for the bromine atom was generally disadvantageous in this series for their activity in vitro. Whereas in the other series, the positive effect in lowering blood pressure in hypertensive animals was observed [25].

Experimental protocols

Melting points were determined on a Gallenkamp apparatus and are uncorrected. Infrared (IR) spectra were obtained with a Perkin Elmer FTIR-1600. The ¹H-NMR spectra were recorded on Varian Gemini 200 or 300 or Bruker-300 spectrometer with tetramethylsilane as internal standard and using CDCl₃ as the solvent; chemical shifts are in δ (ppm) and coupling constants (*J*) are measured in Hz. All compounds were analyzed for C, H and N. Analytical results obtained for these elements were within $\pm 0.4\%$ of the calculated values for the formula shown. All compounds were purified by chromatography column on

silica gel (Merck 70-200 mesh) and used for the subsequent reaction. Thin layer chromatography was performed on silicagel pre-coated plates (Merck, Kieselgel 60 F-254). All reagents were of commercial quality or were purified before use. Organic solvents were purified by standard procedures.

N-Acetylspiro[1,3-benzodioxolo-2,4'-piperidine] and N-acetylspiro[1,3-benzodioxane-2,4'-piperidine] **4a,b**. General procedure

A mixture of N-acetyl-4-piperidone (7.05 g, 50 mmol), methyl orthoformate (6.36 g, 60 mmol), methanol (30 mL), and p-toluenesulfonic acid (50 mg) was gently heated, while the methyl formate formed (bp 34 °C) in the reaction was distilled off through a short vigreux column. The cooled reaction mixture was made basic by addition of few drops of 2 N sodium methoxide, and partitioned between ether (100 mL) and water (100 mL). The organic phase was washed with brine (50 mL) and dried with potassium carbonate. Evaporation of the solvent affords a good yield of dimethyl acetal. Purification was performed by flash chromatography. The crude product thus obtained may be used in the following step without further purification.

A mixture of ketone dimethyl acetal (9.35 g, 50 mmol), 1,2-dihydroxybenzene (6.6 g, 60 mmol), or 2-hydroxybenzyl alcohol (8.9 g, 72 mmol) and benzene (100 mL) was brought reflux and part of the solvent (10 mL) was distilled through a short vigreux column. Heating was discontinued, the temperature was lowered to 60 °C and p-toluenesulfonic acid (50 mg) was added. The distillation of the solvent was slowly continued until pure benzene was being collected. During this operation, addition of benzene was sometimes necessary to prevent the reaction mixture from becoming dry. Triethylamine (1 mL) was added to the cooled reaction mixture, which was then partioned between ether (100 mL) and water (100 mL). The organic phase was washed with water (2 x 50 mL), 10% sodium hydroxide (50 mL), brine (50 mL), dried with potassium carbonate and the solvent evaporated. Purification was performed by flash chromatography (petroleum ether/ethyl acetate 70:30).

Hydrolysis of the acetamide to 5a,b. General procedure

A solution of 4 (1.28 mmol) in ethanol (15 mL) and KOH/water (1.15 g in 5 mL) was heated at reflux for 12 h. Solvents were removed in vacuo, the residue was dissolved in water and extracted with dichloromethane to afford an oil, which was chromatographed on a silica-gel column (ether/dichloromethane 80:20) to give the amine.

General bromination to 5c

To a solution of **5a** chlorhydrate (230 mg, 1.07 mmol) in 6 mL methanol cooled in an ice/salt bath, was added dropwise the NBS (180 mg, 1.01 mmol). The mixture was stirrred for 2 h. The solvent was removed in vacuo, the residue was dissolved in water and extracted with dichloromethane afford an oil, which was chromatographed on a silica gel (hexane/ethyl acetate/ethanol 70:29:1) to give 254 mg of amine **5c**, yield 88%.

Alkylation to 6a-o. General procedure

To 20 mL dry DMF was added the amine (1 mmol), 1.93 g (14 mmol) of anhydrous potassium carbonate, 10 mg KI and then the epoxide (1.5 mmol) or alkylating agent (5 mmol). After stirring at room temperature for 72 h under argon, the

mixture was cooled, filtered and the solvent evaporated to an oil. The oil was dissolved in 15 mL water and 30 mL 3 N hydrochloric acid and was extracted with ether. The water layer was basified with 5 N sodium hydroxide. The alkaline mixture was extracted with dichloromethane and the organic extracts were successively dried and evaporated to dryness, leaving an oil. The oil was chromatographed on a silica-gel column eluting with (ether/dichloromethane 80:20).

Analytical data

N-Acetylspiro[1,3-benzodioxolo-2,4'-piperidine] **4a** C₁₃H₁₅NO₃, n = 0. Yield 65%. Mp 162–163 °C (ethanol). IR (ν cm⁻¹), 1640, 1590, 1210, 1160; ¹H-NMR δ (ppm) (250 MHz, CDCl₃), 2.01 (m, 4H, CH₂), 2.18 (s, 3H, CH₃), 3.62 (t, 2H, J = 5, CH₂N), 3.81 (t, 2H, J = 5, CH₂N), 6.81 (m, 4H, Ar).

N-Acetylspiro[1,3-benzodioxane-2,4'-piperidine] **4b** $C_{14}H_{17}NO_3$, n=1. Yield 84%. Oil. IR (v cm⁻¹), 1645, 1590, 1240, 1100; ¹H-NMR δ (ppm) (250 MHz, CDCl₃), 1.90 (m, 4H, CH₂), 2.16 (s, 3H, CH₃), 3.60 (m, 2H, CH₂N), 3.81 (m, 2H, CH₂N), 4.81 (s, 2H, CH₂O), 7.02 (m, 4H, Ar).

Spiro[1,3-benzodioxolo-2,4'-piperidine] **5a** C₁₁H₁₃NO₂, n = 0. Yield 97%. Mp 244–246 °C (ethanol). IR (ν cm⁻¹), 3300, 1480, 1230, 850; ¹H-NMR, δ (ppm) (250 MHz, CDCl₃), 1.51 (s, 1H, NH), 1.97 (t, J = 6, 4H, CH₂C), 3.01 (t, J = 6, 4H, CH₂N), 6.75 (s, 4H, Ar).

Spiro[1,3-benzodioxane-2,4'-piperidine] **5b** $C_{12}H_{15}NO_2$, n=1. Yield 98%. Oil. Mp 240–242 °C (ether). IR (ν cm⁻¹), 3340, 1480, 1225, 850; ¹H-NMR, δ (ppm) (250 MHz, CDCl₃), 1.83 (m, 4H, CH₂C), 2.46 (s, 1H, NH), 2.98 (t, J=5, 4H, CH₂N), 4.81 (s, 2H, CH₂O), 6.82 (m, 3H, Ar), 7.25 (m, 1H, Ar).

5-Bromo spiro[1,3-benzodioxolo-2,4'piperidine] 5c $C_{11}H_{12}BrNO_3$, $R_1 = Br$, n = 0. Yield 88%. Mp 258–259 °C (ether). IR (v cm⁻¹), 3340, 1490, 1235, 660; ${}^{1}H$ -NMR, δ (ppm) (250 MHz, CDCl₃), 1.93 (t, J = 5, 4H, CH₂C), 2.52 (s, 1H, NH), 3.08 (t, J = 5, 1H, CH₂N), 6.61 (d, J = 8, 1H, H₇-Ar), 6.88 (s, 1H, H₄-Ar), 6.92 (d, J = 8, 1H, H₆-Ar).

N-3-(4-Acetylaminophenoxy)-2-hydroxypropyl spiro[1,3-benzodioxolo-2,4'-piperidine] **6a** C₂₂H₂₆N₂O₅, R₁ = H, n = 0, R₂ = u (see fig 2). Yield 65%. Oil. IR (v, cm⁻¹), 3580, 1680, 1590, 1280; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃) 2.21 (m, 4H, CH₂C), 2.25 (s, 3H, CH₃), 2.63 (d, J = 8, 2H, CH₂N), 2.67 (t, J = 12, 2H, CH₂N), 2.98 (t, J = 12, 2H, CH₂N), 3.99 (d, J = 12, 2H, CH₂O), 4.18 (m, 1H,

CHO), 6.96 (m, 4H, Ar), 7.38 (m, 4H, Ar).

N-3-(1-Naphthoxy)-2-hydroxypropyl spiro[1,3-benzodioxolo-2,4'-piperidine] **6b**

 $C_{24}\bar{H}_{25}NO_4$, R_1 = H, n = 0, R_2 = r (see fig 2). Yield 62%. Oil. IR (v, cm⁻¹), 3590, 1590, 1280, 1100; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.51 (m, 4H, CH₂C), 2.83 (m, 2H, CH₂N), 2.98 (m, 2H, CH₂N), 3.5 (m, 2H, CH₂N), 3.98 (m, 2H, CH₂O), 4,32 (m, 1H, CHO), 6.83 (t, J = 8, 2H, Ar), 7.48 (m, 5H, Ar), 7.89 (m, 2H, Ar), 8.23 (m, 1H, Ar), 8.34 (m, 1H, Ar).

erythro-N-[2-Hydroxy-2(1,4-benzodioxan-2-yl)]ethyl spiro[1,3-benzodioxolo-2,4'-piperidine] **6c**

 $C_{21}H_{23}NO_5$, $R_1 = H$, n = 0, $R_2 = s$ (see fig 2). Yield 43%. Oil. IR (v, cm⁻¹), 3500, 1410, 1270, 1100; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.22 (m, 4H, CH₂C), 2.60 (d, J = 8, 2H,

CH₂N), 2.69 (m, 2H, CH₂N), 3.00 (m, 2H, CH₂N), 3.91–4.20 (m, 4H, CHO), 6.96 (m, 4H, Ar), 7.12 (m, 4H, Ar).

threo-N-[2-Hydroxy-2(1,4-benzodioxan-yl)]ethyl spiro[1,3-benzodioxolo-2,4'-piperidine] **6d**

 $C_{21}H_{23}NO_5$, $R_1 = H$, n = 0, $R_2 = t$ (see fig 2). Yield 80%. Oil. IR (ν, cm⁻¹), 3530, 1480, 1280, 1190; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.12 (m, 4H, CH₂C), 2.54 (dd, J = 5, J = 12, 2H, CH₂N), 2.65 (m, 2H, CH₂N), 2.95 (m, 2H, CH₂N), 3.95 (t, J = 5, 1H, CHO), 4.18 (dt, J = 5, 1H, CHO), 4.35 (m, 2H, CHO), 6.98 (m, 8H, Ar).

N-(1,4-Benzodioxan-2-ylmethyl) spiro[1,3-benzodioxolo-2,4'-piperidine] **6e** $C_{20}H_{21}NO_4$, $R_1=H$, n=0, $R_2=v$ (see fig 2). Yield 72%. Oil. IR (v, cm^{-1}) , 1480, 1230, 1080, 790; ¹H-NMR, δ (ppm)

 (v, cm^{-1}) , 1480, 1230, 1080, 790; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.12 (t, J=4, 4H, CH₂C), 2.85 (m, 6H, CH₂N), 4.01 (dd, J=7, J=9, 2H, CH₂O),4.45 (m, 1H, CHO), 6.85 (m, 8H, Ar).

N-3-(4-Acetylaminophenoxy)-2-hydroxypropyl spiro[1,3-benzo-dioxane-2,4'-piperidine] **6f**

 $C_{23}H_{28}N_2O_5$, R_1 = H, n = 1, R_2 = u (see fig 2). Yield 84%. Oil. IR (v, cm⁻¹), 3570, 1670, 1580, 1400; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.05 (m, 4H, CH₂C), 2.15 (s, 3H, CH₃), 2.62 (m, 4H, CH₂N), 2.85 (m, 2H, CH₂N), 3.99 (d, J = 5, 2H, CH₂O), 4.05 (m, 1H, CHO), 4.80 (s, 2H, CH₂O), 7.06 (m, 8H, Ar).

N-3-(1-Naphthoxy)-2-hydroxypropyl spiro[1,3-benzodioxane-2,4'-piperidine] **6g**

 $C_{25}H_{27}NO_4$, R_1 = H, n = 1, R_2 = r (see fig 2). Yield 58%. Oil. IR (V, cm⁻¹), 3590, 1590, 1450, 1260; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.02 (m, 4H, CH₂C), 2.62 (m, 2H, CH₂N), 2.73 (d, J = 5, 2H, CH₂N), 2.83 (m, 2H, CH₂N), 4.32 (m, 3H, CHO), 4.73 (s, 2H, CH₂O), 6.98 (m, 4H, Ar), 7.21 (m, 1H, Ar), 7.52 (m, 4H, Ar), 7.83 (m, 1H, Ar), 8.43 (m, 1H, Ar).

erythro-N-[2-Hydroxy-2-(1,4-benzodioxan-2-yl)]ethyl spiro [1,3-benzodioxane-2,4'-piperidine] **6h** C₂₂H₂₅NO₅, R₁ = H, n = 1, R₂ = s (see fig 2). Yield 45%. Oil. IR (ν, cm⁻¹), 3560, 1590, 1210, 1190; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.01 (m, 4H, CH₂C), 2.58 (dd, J = 5, J = 11, 2H, CH₂N), 2.65 (m, 2H, CH₂N), 2.97 (m, 2H, CH₂N), 3.98 (t, J = 3, 1H, CHO), 4.23 (t, J = 3, 1H, CHO), 4.37 (m, 2H, CH₂O), 4.79 (s, 2H, CH₂O), 6.99 (m, 8H, Ar).

threo-N-[2-Hydroxy-2-(1,4-benzodioxan-2-yl)]ethyl spiro [1,3-benzodioxane-2,4'-piperidine] 6i $C_{22}H_{25}NO_5,1=H,\ n=1,\ R_2=t\ (see\ fig\ 2).$ Yield 33%.Oil. IR (v, cm⁻¹), 3530, 1490, 1260, 1105; ¹H-NMR, δ (ppm)

(8, CH, CH₂), 3330, 1430, 1200, 1103, 11-141HH, C (ppln) (300 MHz, CDCl₃), 1.98 (m, 4H, CH₂C), 2.51 (m, 2H, CH₂N), 2.88 (m, 2H, CH₂N), 2.89 (dd, J = 5, J = 11, 2H, CH₂N), 3.92 (m, 1H, CHO), 4.21 (m, 1H, CHO), 4.42 (m, 2H, CHO), 4.87 (s, 2H, CH₂O), 6.95 (m, 7H, Ar), 7.21 (m, 1H, Ar).

N-(1,4-Benzodioxan-2-yl methyl)spiro[1,3-benzodioxane-2,4'-piperidine] 6j

 $C_{21}H_{23}NO_4$, R_1 = H, n = 1, R_2 = v (see fig 2). Yield 58%. Oil. IR (v, cm⁻¹), 1480, 1235, 1075, 790; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 1.95 (m, 4H, CH₂C), 2.63 (m, 6H, CH₂N), 4.01 (m, 2H, CH₂O), 4.38 (m, 1H, CHO), 4.87 (s, 2H, CH₂O), 6.94 (m, 7H, Ar), 7.06 (m, 1H, Ar).

5-Bromo-N-3-(4-acetylaminophenoxy)-2-hydroxypropyl spiro[1,3-benzodioxolo-2,4'-piperidine] $\boldsymbol{6k}$ C₂₂H₂₅BrN₂O₅, R₁ = Br, n = 0, R₂ = u (see fig 2). Yield 90%. Oil. IR (ν , cm⁻¹), 3510, 1490, 1250, 790; ¹H-NMR, δ (ppm)

(300 MHz, CDCl₃), 2.01 (m, 4H, CH₂C), 2.21 (s, 3H, CH₃), 2.61 (m, 2H, CH₂N), 2.73 (m, 2H, CH₂N), 2.83 (m, 2H, CH₂N), 3.98 (d, J = 6, 2H, CH₂O), 4.12 (m, 1H, CHO), 7.06 (m, 7H, Ar).

5-Bromo-N-3(1-naphthoxy)-2-hydroxypropyl spiro[1,3-benzo-dioxolo-2,4'-piperidine] **61**

 $C_{24}H_{24}BrN_2O_4$, $R_1 = Br$, n = 0, $R_2 = r$ (see fig 2). Yield 43%. Oil. IR (v, cm⁻¹), 3490, 1480, 1100, 1080; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.03 (m, 4H, CH₂C), 2.64 (m, 4H, CH₂N), 2.78 (m, 2H, CH₂N), 4.35 (m, 3H, CHO), 6.56 (d, J = 6, 1H, Ar), 6.83 (m, 2H, Ar), 7.47 (m, 5H, Ar), 7.78 (m, 1H, Ar), 8.42 (m, 1H, Ar).

erythro-5-Bromo-N-[2-hydroxy-2-(1,4-benzodioxan-2-yl)]ethyl spiro [1,3-benzodioxolo-2,4'-piperidine] **6m**

 $C_{21}H_{22}BrNO_5$, $R_1 = Br$, n = 0, $R_2 = s$ (see fig 2). Yield 62%. Oil. IR (v, cm⁻¹), 3490, 1485, 1232, 1070; ¹H-NMR, δ (ppm) (200 MHz, CDCl₃), 2.00 (m, 4H, CH₂C), 2.51 (m, 2H, CH₂N), 2.88 (m, 2H, CH₂N), 2.89 (m, 2H, CH₂N), 3.89 (m, 1H, CHO), 4.12 (dt, J = 3, 1H, CHO), 4.23 (m, 2H, CH₂O), 6.66 (d, J = 6, 1H, Ar), 6.97 (m, 6H, Ar).

threo-5-Bromo-N-[2-hydroxy-2-(1,4-benzodioxan-2-yl)]ethyl spiro [1,3-benzodioxolo-2,4'-piperidine] **6n**

3ριο 11,3-0ch20thoxolo-2,3-peperhatary M C₂₁H₂₂BrNO₅, R₁ = Br, n = 0, R₂ = t (see fig 2). Yield 71%. Oil. IR (v, cm⁻¹), 3500, 1490, 1200, 1100; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.02 (m, 4H, CH₂C), 2.52 (m, 2H, CH₂N), 2.63 (m, 2H, CH₂N), 2.84 (m, 2H, CH₂N), 3.98 (m, 1H, CHO), 4.12 (m, 1H, CHO), 4.45 (m, 2H, CHO), 6.62 (d, J = 7, 1H, Ar), 6.98 (m, 6H, Ar).

5-Bromo-N-(1,4-benzodioxan-2-ylmethyl) spiro[1,3-benzodioxolo-2,4'-piperidine] **60**

C₂₀H₂₀BrNO₄, R₁ = Br, n = 0, R₂ = v (see fig 2). Yield 70%. Oil. IR (v, cm⁻¹), 1475, 1220, 1080, 790; ¹H-NMR, δ (ppm) (300 MHz, CDCl₃), 2.10 (t, J = 5, 4H, CH₂C), 2.63 (dd, $J_1 = 5$, $J_2 = 11$, 2H, CH₂N), 2.75 (m, 4H, CH₂N), 4.01 (dd, $J_1 = 5$, $J_2 = 11$, 1H, CHO), 4.26 (m, 1H, CHO), 4.46 (m, 1H, CHO), 6.63 (d, J = 6, 1H, Ar), 6.87 (m, 6H, Ar).

2-Iodomethyl-1,4-benzodioxan 7

Diethyl azodicarboxylate (DEAD, 2 mmol) was added dropwise at 0 °C to a stirred dry toluene suspension containing the

alcohol (1 mmol), triphenylphosphine (2 mmol) and ZnI_2 (0.75 mmol). After the reaction was complete, purification was effected by direct silica-gel column chromatography (hexane/ethyl acetate 90:10). Yield 85%. ¹H-NMR was identical that the spectral data described in reference [14].

References

- 1 Kosley RW (1984) US Patent 4, 452 802; Chem Abstr (1984) 101, 130 600k
- 2 Bauer VJ, Duffy BJ, Hoffman D et al (1976) J Med Chem 19, 1315-1324
- 3 Allen RC, Bauer VJ, Kosley RW et al (1978) J Med Chem 21, 1149-1154
- 4 Klioze SS, Allen JC, Wilker JC, Woodward DL (1980) J Med Chem 23, 677-679
- 5 Klioze SS, Novick WJ (1978) J Med Chem 21, 400-402
- 6 Main BG, Tucker H (1985) In: Progress in Medicinal Chemistry 22 (Ellis GP, West GB, eds) Elsevier Science Publishers (Biomedical Division), Amsterdam, 122–158
- 7 Crowther AF, Smith LH (1968) J Med Chem 11, 1009-1013
- 8 Zhang YL, Trapsha A, McPhail AT, Lee KH (1994) J Med Chem 37, 1460-
- 9 Cole ER, Crank G, Minh H (1980) Aust J Chem 33, 675-680
- 10 Napolitano E, Fiaschi R, Mastrorilli E (1986) Synthesis 122-125
- 11 Mauleón D, Pujol MD, Rosell G (1988) Eur J Med Chem 23, 421-426
- 12 Gschwend HW, Huebner CF (1980) US Patent 4, 212, 808; Chem Abstr (1980) 93, 220 757n
- 13 Koo J, Avakian S, Martin J (1955) J Am Chem Soc 77, 5373-5379
- 14 Ruiz N, Rollin P (1989) Tetrahedron Lett 30, 1637-1640
- 15 Mitsunobu O (1981) Synthesis, 1-28
- 16 Mauleón D, Pujol MD, Rosell G (1988) J Med Chem 31, 2122-2126
- 17 Van Rossum JM (1963) Arch Int Pharmacodyn 43, 315-320
- 18 Main BG, Tucker H (1985) In: Medicinal Chemistry Vol 22, Academic Press, London, 69–92
- 19 Clark RD, Michel AD, Whiting RL (1986) In: Progress in Medicinal Chemistry Vol 23 (Ellis GP, West GB, eds) Elsevier, Amsterdam, 1-40
- 20 Ing HR, Ormerad WE (1952) *J Pharm Pharmacol* 4, 21–26
- 21 Lohmann D, Lehmann D, Morgenstern E, Faust G (1990) Pharmazie 45, 401–404
- 22 Uloth RH, Kirk JR, Gould WA, Larsen AA (1966) J Med Chem 9, 88-97
- 23 Howe R (1969) J Med Chem 12, 642-646
- 24 Tucker H (1981) J Med Chem 24, 1364-1368
- 25 Kosley RW, Cherill RJ (1984) US Patent 4, 452 802; Chem Abstr (1984) 101, 130 600 k